

# **General Chemistry**

Programme course

6.0 credits

Allmän kemi

8BKG13

Valid from: 2019 Autumn semester

**Determined by** The Board for First and Second Cycle Programmes at the Faculty of Medicine and Health Sciences

Date determined 2017-08-22

#### Main field of study Chemistry

**Course level** 

First cycle

### Advancement level

G1X

### Course offered for

• Experimental and Industrial Biomedicine

### Entry requirements

General entry requirements for undergraduate studies and English corresponding to the level of English in Swedish upper secondary education (English 6) and Chemistry, Mathematics and Biology corresponding to the level in Swedish upper

Chemistry, Mathematics and Biology corresponding to the level in Swedish upper secondary education (Chemistry 2, Mathematic 4 and Biology 2) Exemption from Swedish 3



### Intended learning outcomes

Knowledge and understanding

On completion of the course, the student shall be able to:

- Describe the structures of atoms and molecules
- Discuss different types of chemical bonding and the relationship between bonding and state of aggregation
- Explain the meaning of chemical equilibrium, and its applications on equilibria of acids and bases along with solubility equilibria
- Explain the energetics and kinetics of chemical reactions
- Describe the laws of thermodynamics, in particular their application on chemical systems

Skills and abilities

On completion of the course, the student shall be able to:

- Perform chemical laboratory techniques and theoretical analysis of experimental data and present this both orally and in writing
- Perform stoichiometric calculations and apply them in laboratory work within chemistry

Judgement ability and approach

On completion of the course, the student shall be able to:

• Critical appraise laboratory work on the basis of aspects relating to safety

### Course content

The course involves the study of general chemistry, including the structure of atoms and molecules, stoichiometry, chemical equilibrium such as acid-base equilibrium and solubility equilibrium. The course also includes chemical bonding, kinetics of chemical reactions, the three laws of thermodynamics and the concepts enthalpy, entropy and free energy. Furthermore, basic laboratory techniques and safety are introduced. Basic knowledge in the field of general chemistry is prepares the student for advanced courses in biochemistry. The course covers general chemistry, inorganic chemistry and thermodynamics



### Teaching and working methods

At the Faculty of Medicine and Health Sciences student centred and problem based learning make up the foundation of the teaching. The student takes responsibility for, studies and researches current content of the courses and study programme. The methods of the course work challenge the students to independently formulate questions for learning, to seek knowledge and in dialogue with others judge and evaluate achieved knowledge. Students in the Bachelor's Programme in Experimental and Industrial Biomedicine work together in groups based based on reality based and course-related biomedical issues to apply their knowledges, develop their own learning, contribute to the fellow student's learning and to practice cooperation. Throughout the study programme theory is integrated with practical modules. The course methods and integration modules stimulates and support the student's ability to apply their knowledge and professional competence.

Working methods used on this course are lectures, classes, seminars and laboratory exercises.

### Examination

The examination consists of one individual written exam. In addition, active participation in compulsory elements is required in order to pass the course. Compulsory elements include seminars, laboratory sessions, reports and assignments.

The written exam may be performed an unlimited number of times by those students who have not achieved a passing grade.

Point of time for retake examination must normally be announced no later than the time of the regular examination. The extent of the retake examination must be the same as the regular examination.

CHANGE OF EXAMINER

A student who has obtained a failing grade twice for a course or part of a course is, after request, entitled to be appointed another examiner, unless there are special reasons to the contrary.

#### APPLICATION FOR EXAMINATION / WRITTEN EXAM

Instructions on how to apply for examinations are given prior the beginning of each course.

### Grades

Four-grade scale, digits, U, 3, 4, 5

### **Course literature**

A literature reference list must be set no later than two months before the course begins by the programme committee for the Bachelor's Programme in Experimental and Industrial Biomedicine. There is no compulsory course literature.



### Other information

Planning and implementation of the course is to be based on the wording in the course syllabus. A course evaluation is compulsory for each course and should include how the course is in agreement with the course syllabus. The course coordinator will analyse the course evaluation and propose appropriate development of the course. The analysis and proposal will be returned to the students, the Director of Studies, and as needed to the Education board, if related to general development and improvement.

The course is carried out in such a way that knowledge of gender, gender identity/expression, ethnicity, religion or other belief system, disability, sexual orientation and age is addressed, highlighted and communicated as part of the programme.

If the course is cancelled or undergoes major changes, examination is normally offered under this course syllabus, at a total of three occasions,

within/in connection to the two following semesters, of which one will in close proximity to the first examination.

## Department

Medicinska fakulteten

