

# Analytical Chemistry T

Programme course

6 credits

Analytisk kemi T

NKEB05

Valid from: 2017 Spring semester

**Determined by** Board of Studies for Chemistry, Biology and Biotechnology

**Date determined** 2017-01-25

# Main field of study

Chemical Engineering, Chemistry

#### Course level

First cycle

#### Advancement level

G<sub>1</sub>X

# Course offered for

- Chemistry, Bachelor's Programme
- Chemical Analysis Engineering, B Sc in Engineering

# **Entry requirements**

Note: Admission requirements for non-programme students usually also include admission requirements for the programme and threshold requirements for progression within the programme, or corresponding.

# **Prerequisites**

General Chemistry, Organic Chemistry, Inorganic Chemistry and Calculation tools for chemistry students

# Intended learning outcomes

The aim of the course is to give fundamental theoretical and practical knowledge in the field of volumetric analysis (titrations). After completing this course the student should be able to:

- Explain the chemical principles of volumetric analyses.
- Plan, perform and evaluate chemical analyses based on titrations.
- Write reaction equations and perform calculations for volumetric analyses.
- Calculate and analyse titration curves.
- Construct logaritmic pH diagrams and use these to describe acid-base titration equilibria.

#### Course content

General aspects of analytical chemistry. Acid-base equilibria and pH graphs (logarithmic diagrams). Quantitative chemical analysis based on volumetric methods: acid-base, redox, complexometric and precipitation titrations – theory and applied laborations where authentic samples are analysed. Titrations with indicators and potentiometric titrations. Experimental errors.



# Teaching and working methods

The course consists of lectures, lessons, laborations, computer exercises and short written tests. The laborations and computer exercises are compulsory.

#### **Examination**

LAB1	Laboratory work	3 credits	U, G
TEN <sub>1</sub>	Written examination	3 credits	U, 3, 4, 5

#### Grades

Four-grade scale, LiU, U, 3, 4, 5

#### Other information

Supplementary courses: Analytical chemistry S, Analytical chemistry K and Organic Analytical Chemistry

# Department

Institutionen för fysik, kemi och biologi

# Director of Studies or equivalent

Magdalena Svensson

#### **Examiner**

Johan Dahlén

## Course website and other links

# **Education components**

Preliminary scheduled hours: 94 h Recommended self-study hours: 66 h

## Course literature

Harris D.C., Quantitative Chemical Analysis, 8th ed., Freeman, 2010.



#### **Common rules**

Regulations (apply to LiU in its entirety)

The university is a government agency whose operations are regulated by legislation and ordinances, which include the Higher Education Act and the Higher Education Ordinance. In addition to legislation and ordinances, operations are subject to several policy documents. The Linköping University rule book collects currently valid decisions of a regulatory nature taken by the university board, the vice-chancellor and faculty/department boards.

LiU's rule book for education at first-cycle and second-cycle levels is available at http://styrdokument.liu.se/Regelsamling/Innehall/Utbildning\_pa\_grund\_och\_avancerad\_niva.

